Research Article

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ONE-POT SYNTHESIS AND ANTIMICROBIAL ACTIVITY OF 2-AMINO-5-ARYL-5H-THIAZOLO [4,3-b]-1,3,4-THIADIAZOLES

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ABSTRACT

A series of 2-amino-5-aryl-5H-thiazolo[4,3-b]-l,3,4-thiadiazoles were synthesized by using aromatic aldehydes, thioglycolic acid and thiosemicarbazide. Equimolar mixtures of aromatic aldehydes with thioglycolic acid and thiosemicarbazide in H₂SO₄ transform into 2-amino-5-aryl-5H-thiazolo[4,3-b]-l,3,4-thiadiazoles. Structures of all synthesized compounds were confirmed by FTIR, ¹H NMR and mass spectral data. Fungicidal activity against two fungi Aspergillus niger and Candida albicans and bactericidal activity against two gram +ve bacteria Staphylococcus aureus, Escherichia fecalis and two gram –ve bacteria Escheridhia coli, Klebsiella pneumonia and found to be active against these microorganisms.

KEYWORDS: Aromatic aldehydes, thioglycolic acid, thiosemicarbazide, 2-aryl-(4-oxothiazolidin-3-yl)thiourea, cyclodehydration, 2-amino-5-aryl-5H-thiazolo[4,3-b]-1,3,4-thiadiazoles, Antibacterial activity, Antifungal activity.

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INTRODUCTION

The resistance towards available drugs is rapidly becoming a major worldwide problem. The need to design new compounds to deal with this resistance has become one of the most important areas of research today. Thiadiazole is a versatile moiety that exhibits a wide variety of biological activities. Thiadiazole moiety acts as "hydrogen binding domain" and "two-electron donor system". It also acts as a constrained pharmacophore. Many drugs containing thiadiazole nucleus are available in the market such as acetazolamide, methazolamide, sulfamethazole, etc. Thiadiazole can act as the bio-isosteric replacement of the thiazole moiety. So it acts like third and fourth generation cephalosporins, hence can be used in antibiotic preparations. Thiadiazole is a 5-membered ring system containing two nitrogen and one sulphur atom. They occur in nature in four isomeric forms viz. 1,2,3thiadiazole; 1,2,5-thiadiazole; 1,2,4-thiadiazole and 1,3,4-thiadiazole. The 1,3,4-thiadiazole isomer of thiadiazole series and its dihydro-derivatives provide a

bulk of literature on thiadiazole. A glance at the standard reference work shows that more work has been carried out on the 1.3.4-thiadiazole than all other isomers combined. Members of this ring system have found their way into such diverse application as pharmaceuticals, oxidation inhibitors, cyanine dyes, & metal complexing agents. In past decades, thiadiazoles have proved their potential in development of pharmaceutically important organic compounds both of natural and synthetic origin¹. Thiadiazole analogs deal with a variety of bioactivities antitumor²⁻³, anti-HIV⁴, antimicrobial⁵⁻⁶, viz. anticonvulsant⁷, antitubercular⁸, antiprotozoal⁹, anti-inflammatory¹⁰. Literature is enriched with lot of work on synthesis of potent substituted thiadiazole derivatives with diverse pharmacological activities¹¹⁻¹² but only few reports have been received on peptide coupling of thiadiazoles. Thus keeping in view the biological potency of thiadiazole derivatives a series of 2-amino-5with an anticipation to get potent agents of more therapeutic efficacy with lesser adverse effects. The wide range of



therapeutic values of thiadiazoles prompted us to synthesize the title compounds and screen them for their antimicrobial activities.

Recently, N-substituted-2-amino-5-arvl-5H-thiazolo[4,3b]-1,3,4-thiadiazoles, active against pathogenic fungi of agricultural crops, have become substances, among other 1,3,4-thiadiazoles, that have drawn the attention of researchers¹³⁻¹⁵. The only known method giving rise to that heterocyclic system is based on cyclodehydration of (4-oxothiazolidine-3-yl)thioureas (in concentrated H₂SO₄ medium). which in turn are formed upon cyclodehydration of the products of addition of thioglycolic acid to the azomethine fragment of arylthiosemicarbazones,

MATERIALS AND METHODS

Analytical grade solvents and commercially available reagents were used without further purification. The column chromatography was carried out over silica gel (60-120 mesh), purchased from Sisco Research Laboratories Pvt Ltd. Melting points were determined in Prog, melting point apparatus Servewell DBK, Instruments Pvt Ltd. IR spectra in KBr disk were recorded from 4000 to 400 cm⁻¹ on Shimadzu FT-IR spectrometer. ¹H NMR spectra were recorded on 400-MHz and 500-MHz Bruker spectrometer in CDCl₃ using tetramethylsilane (TMS) as an internal standard. Chemical shifts are given in δ relative to TMS, the coupling constants are given in Hz. Mass spectra were recorded using Agilent 1100 MSD spectrometer in electro spray mode.

General procedure of synthesis of compounds 5a--5j. An aromatic aldehyde (0.02 mole) and thioglycolic acid (0.02 mole) were mixed, and after 10--15 min. 0.022 mole of thiosemicarbazide was added; then 10 mL of concentrated H_2SO_4 was added in portions upon cooling. The mixture was homogenized and left for 18--24 hours at -20 °C. The reaction mass was treated with 30--50 g ice, the precipitated solid was decanted, water was added, and the obtained suspension was neutralized with 40% NaOH until a weak alkaline reaction. Compounds 3a--j were recrystallized from aqueous dioxane solution.

2-Amino-5-phenyl-5H-thiazolo[4,3-b]-1,3,4-

thiadiazole (5a). yield 87%, m.p. $156-158^{\circ}$ C, yellowish crystals. IR spectrum (v/cm⁻¹): 3472 (-NH₂); 3155(-CH aromatic); 2821 (-CH of thiazole); ¹H NMR spectrum (δ , ppm): 11.42 (s, CH); 8.19 (s, CH); 7.36-7.41 (m, Ph). MS spectrum, *m/z*: 236 [M+1]⁺.

2-Amino-5-(4-methylphenyl)-5H-thiazolo[4,3-b]-1,3,4thiadiazole (5b). yield 88%, m.p. $173-174^{\circ}$ C, yellowish crystals. IR spectrum (v/cm⁻¹): 3402 (-NH₂); 3155 (-CH aromatic); 2989 (-CH aliphatic); 2800 (-CH of thiazole); ¹H NMR spectrum (δ , ppm): 10.20 (s, CH); 6.56 (s, CH); 7.18-7.55 (m, Ph); 2.37 (s, Me). MS spectrum, m/z: 250 $[M+1]^+$.

2-Amino-5-(4-hydroxyphenyl)-5H-thiazolo[4,3-b]-

1,3,4-thiadiazole (5c). yield 85%, m.p. 226-227°C, yellow crystals. IR spectrum (v/cm⁻¹): 3471 (-NH₂); 3360 (-CH aromatic; 2880 (-CH of thiazole); ¹H NMR spectrum (δ , ppm): 11.14 (s, CH); 7.98 (s, CH); 5.55(s, OH) 7.08-7.74 (m, Ph). MS spectrum, *m/z*: 252 [M+1]⁺.

2-Amino-5-(4-nitrophenyl)-5H-thiazolo[4,3-b]-1,3,4thiadiazole (5d). yield 87.3%, m.p. 158-160 0 C, yellowish crystals. IR spectrum (v/cm⁻¹): 3491 (-NH₂); 3365 (-CH aromatic); 2995 (-CH of thiazole); ¹H NMR spectrum (δ , ppm): 11.14 (s, CH); 7.98 (s, CH); 7.05-7.70 (m, Ph). MS spectrum, *m/z*: 281 [M+1]⁺.

2-Amino-5-(4-dimethylaminophenyl)-5H-thiazolo[4,3b]-1,3,4-thiadiazole (5e). yield 77%, m.p. 209-210^oC, yellowish crystals. IR spectrum (v/cm⁻¹): 3440 (-NH₂); 3115 (-CH aromatic); 2960 (-CH aliphatic); 2880 (-CH of thiazole); ¹H NMR spectrum (δ , ppm): 10.20 (s, CH); 8.27 (s, CH); 7.16-7.30 (m, Ph); 3.77 (s, Me). MS spectrum, *m/z*: 279 [M+1]⁺.

2-Amino-5-(2-chlorophenyl)-5H-thiazolo[4,3-b]-1,3,4thiadiazole (5f). yield 82%, m.p. $214-215^{0}$ C, yellow crystals. IR spectrum (v/cm⁻¹): 3410 (-NH₂); 3115 (-CH aromatic); 2800 (-CH of thiazole); ¹H NMR spectrum (δ , ppm): 11.14 (s, CH); 7.96 (s, CH); 7.08-7.74 (m, Ph). MS spectrum, *m/z*: 271 [M+1]⁺.

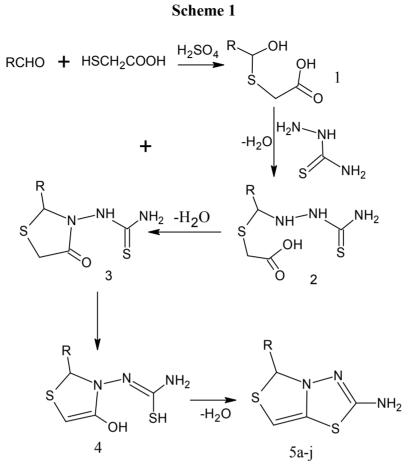
2-Amino-5-(4-chlorophenyl)-5H-thiazolo[4,3-b]-1,3,4-thiadiazole (5g). yield 91%, m.p. 207-208°C, yellowish crystals. IR spectrum (v/cm⁻¹): 3449 (-NH₂); 3282 (-CH aromatic); 2995 (-CH of thiazole); ¹H NMR spectrum (δ , ppm): 9.51 (s, CH); 6.4 (s, CH); 7.28-7.72 (m, Ph). MS spectrum, *m/z*: 271 [M+1]⁺.

2-Amino-5-(2,4-dichlorophenyl)-5H-thiazolo[4,3-b]-1,3,4-thiadiazole (5h). yield 89%, m.p. 241-242 $^{\circ}$ C, Brownish yellow crystals. IR spectrum (v/cm⁻¹): 3456 (-NH₂); 3260 (-CH aromatic); 3020 (-CH of thiazole); ¹H

NH₂); 3260 (-CH aromatic); 3020 (-CH of thiazole); ¹H NMR spectrum (δ , ppm): 9.25 (s, CH); 7.32-7.76 (m, Ph). MS spectrum, *m/z*: 305 [M+1]⁺.

2-Amino-5-(2-methoxyphenyl)-5H-thiazolo[4,3-b]-1,3,4-thiadiazole (5i). yield 69%, m.p. 168-169°C, yellowish crystals. IR spectrum (v/cm^{-1}): 3406 (-NH₂); 3288 (-CH aromatic); 3100 (-CH aliphatic); 2800 (-CH of thiazole); ¹H NMR spectrum (δ , ppm): 9.79 (s, CH); 6.86 (s, CH); 7.14-7.79 (m, Ph),3.73 (s, OCH₃). MS spectrum, *m/z*: 266 [M+1]⁺.

2-Amino-5-(3-methoxyphenyl)-5H-thiazolo[4,3-b]-1,3,4-thiadiazole (5j). yield 72%, m.p. 212-213°C, yellowish crystals. IR spectrum (v/cm^{-1}): 3417 (-NH₂); 3250 (-CH aromatic); 3000 (-CH aliphatic); 2800 (-CH of thiazole); ¹H NMR spectrum (δ , ppm): 9.55 (s, CH); 6.92 (s, CH); 7.12-7.75 (m, Ph); 3.77 (s, OCH₃). MS spectrum, m/z: 265 [M+1]⁺.



 $R = Ph (a), 4-MeC_6H_4 (b), 4-HOC_6H_4 (c), 4-O_2NC_6H_4 (d), 4-Me_2NC_6H_4 (e),$ $2-ClC_6H_4 (f), 4-ClC_6H_4 (g), 2,4-Cl_2C_6H_3 (h), 2-OMeC_6H_4 (i), 3-OMeC_6H_4 (j)$

RESULTS AND DISCUSSION Chemistry

The synthesis of the finished products from aromatic aldehydes, thiosemicarbazide, and thioglycolic acid involves three steps. We followed the Shukurov et al¹⁶ procedure for one-pot synthesis of 2-amino-5-aryl-5 Hthiazolo[4,3-b]-1,3,4-thiadiazoles (5a-5j) (Scheme 1) from equimolar quantities of an aromatic aldehyde, thioglycolic acid, and thiosemicarbazide. The interaction between equimolar quantities of aromatic aldehydes and thioglycolic acid proceeds with heat liberation and probably results in semithioacetals of thioglycolic acid (1), which further react with thiosemicarbazide to give functionalized N,S-acetals (2). Thioureas 3 are obtained as a result of cyclodehydration of the latter in concentrated H₂SO₄ medium; they transform into compounds 5a-5j through intermediates 4. The structure of compounds 5a-5j was confirmed by IR, ¹H NMR and Mass spectroscopy. The IR spectra of these compounds have no absorption band in the region of 1680-1630 cm⁻ ¹, characteristic of the stretching vibrations of the

carbonyl group of the amide fragment in compounds of type 3, which confirms the hydrothiazolo[4,3-b]-1,3,4thiadiazole structure. There are two absorption bands in the region $3470-3400 \text{ cm}^{-1}$ and $3250-3220 \text{ cm}^{-1}$ that might be interpreted as asymmetric and symmetric stretching vibrations of the amino group. Four peaks, due to the CH-stretching vibrations of the aromatic ring, were recorded in the interval 3180-2820 cm⁻¹. An intense absorption band is observed in the spectra at 1960 cm⁻¹, which can be assigned to the C-C normal vibrations of the aromatic ring. A group of the absorption bands in the 1600-500 cm⁻¹ region is likely to be associated with the 5-phenyl-5H-thiazolo[4,3-b]-1,3,4-thiadiazole system. The methine proton signals and those of the proton in the 5H-position of the thiazole ring are detected in the ¹H NMR spectra at 7.96--8.3 ppm. The resonance lines of the phenyl ring protons are observed at 7.08--8.27 ppm. **Pharmacological Studies**

Antibacterial and Antifungal activity

Compounds 5a-5j were screened for antibacterial and antifungal activities using the Disc Diffusion method¹⁷

by measuring the zone of inhibiton. A 24 h culture of bacterial strains of *S. aureus*, *E. fecalis*, *K.pneumonia* and *E. coli* were cultivated in Brain heart infusion agar medium and the fungal strains of *A. niger*, and *C. albicans* were cultivated in Sabouraud agar medium respectively. All the compounds were tested at different concentration level. Dimethyl formamide was used as a solvent and as control. Ciprofloxacin and Fluconazole were used as a standard for comparison of the results. The diameter of zone of inhibition was measured in millimeter(mm) after 24h incubation at 37^{0} C. The Antimicrobial study results are tabulated in **Table 1**.

CONCLUSION

We prepared some thiadiazoles 5a-5j in good yields from aromatic aldehvdes, thioglycolic acid and thiosemicarbazide. Equimolar mixtures of aromatic aldehydes with thioglycolic acid and thiosemicarbazide transform into 2-amino-5-aryl-5Hin H_2SO_4 thiazolo[4,3-b]-l,3,4-thiadiazoles and many compounds from this series showed moderate to good antimicrobial activity at low concentrations. The biological profiles of these new generation of thiadiazoles would represent a fruitful matrix for further development of better antimicrobial agents.

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Table 1: Antibacterial and Antifung	al activity of synthesized compounds
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	MIC in μ g/ml and zone of Inhibition in mm					
Compounds	Gram+ve bacteria		Gram -ve bacteria		Fungi tested	
	S. aureus	E. fecalis	K.pneumonia	E. coli	A. niger	C. albicans
5a	50 (8-12)	50 (8-12)	25 (8-12)	50 (8-12)	50 (7-11)	25 (6-10)
5b	50 (6-10)	75 (7-11)	50 (6-10)	50 (6-10)	50 (6-10)	50 (7-11)
5c	75 (12-16)	75 (8-12)	50 (8-12)	75 (8-12)	75 (8-12)	25 (6-10)
5d	50 (7-11)	75 (6-10)	50 (6-10)	75 (6-10)	50 (6-10)	50 (8-12)
5e	75 (11-15)	75 (6-10)	50 (9-13)	75 (8-12)	50 (6-10)	50 (6-10)
5f	50 (6-10)	75 (8-12)	50 (8-12)	75 (8-12)	75 (8-12)	75 (10-14)
5g	75 (8-12)	75 (8-12)	75 (6-10)	75 (6-10)	75 (7-11)	75 (11-15)
5h	75 (9-13)	75 (10-14)	75 (8-12)	75 (8-12)	75 (9-13)	50 (8-12)
5i	75 (6-10)	75 (8-12)	75 (8-12)	75 (6-10)	75 (8-12)	50 (6-10)
5j	75 (8-12)	75 (6-10)	75 (6-10)	75 (8-12)	75 (6-10)	50 (8-12)
DMF (Control)	-	-	-	-	-	-
Ciprofloxacin (Standard)	10 (24-28)	10 (25-29)	10 (28-32)	10 (30-34)	-	-
Fluconazole (Standard)	-	-	-	-	30 (24-28)	30 (22-26)

The MIC values were evaluated at concentration range, 5-75 μ g/ml. The figure in the table showed the value in μ g/ml and the corresponding zone of inhibition in millimeter(mm).

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